

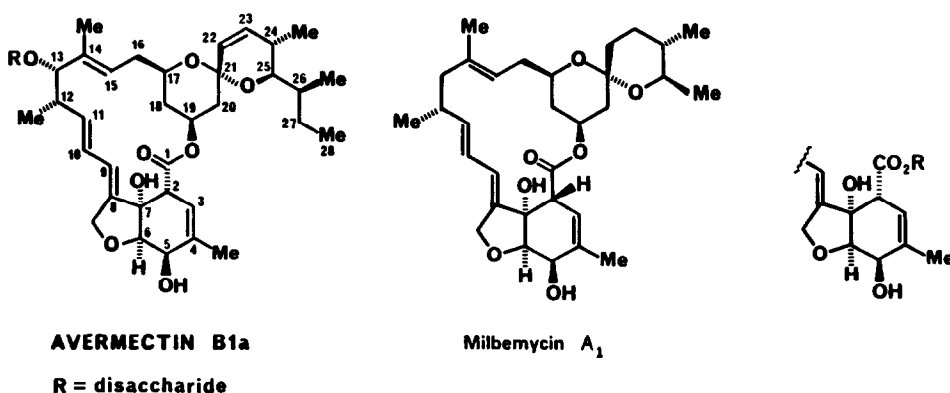
**A NOVEL SYNTHETIC ROUTE TO THE HEXAHYDROBENZOFURAN  
SUBUNIT OF THE AVERMECTINS AND MILBEMYCINS**

Stephen Hanessian\*, Pierre Beaulieu and Daniel Dubé  
Department of Chemistry, Université de Montréal  
Montréal, Québec, Canada H3C 3J7

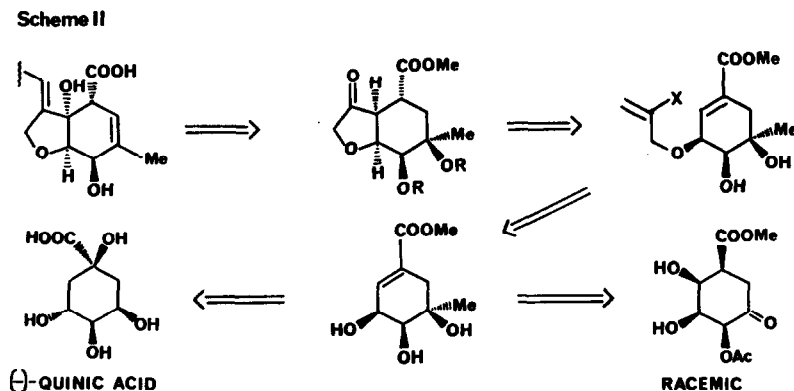
**Summary** - Radical-induced intramolecular Michael cyclization of a bromovinyl-type appendage onto a functionalized cyclohexene carboxylic acid derivative produces the corresponding oxahydrindanes which can be transformed into oxahydrindenes (hexahydrobenzofurans) related to the title compounds. (-)-Quinic acid is a useful optically active precursor.

The recent discovery of the potent classes of anthelmintic agents such as the avermectins<sup>1</sup> and the milbemycins<sup>2</sup> have fostered a great deal of research activities on several fronts. Already a number of total syntheses in the milbemycin area have been reported,<sup>3</sup> and a synthesis of avermectin B<sub>1a</sub> was recently reported from our laboratory.<sup>4</sup> These "megastructures" present a number of synthetic challenges, notably with regard to the hexahydrobenzofuran (oxahydrindene) subunit. The response from the community of synthetic organic chemists has been impressive, with the announcement of no less than four different approaches.<sup>5-8</sup> Three of the four routes<sup>6-8</sup> have actually produced derivatives of the intended bicyclic subunit in which the ester is reduced to an alcohol. This is understandable because of the possibility of  $\beta$ -elimination and subsequent aromatization when working with the acid or ester form (Scheme 1).

Scheme 1



We describe herein a strategically novel and operationally different approach to the synthesis of the hexahydrobenzofuran subunit of avermectin B<sub>1a</sub> and some of the milbemycins, based on the retrosynthetic analysis illustrated in Scheme II. Two of the critical bond-forming steps involved a free-radical mediated intramolecular ring closure, and an oxidative acetoxylation at a ring juncture.

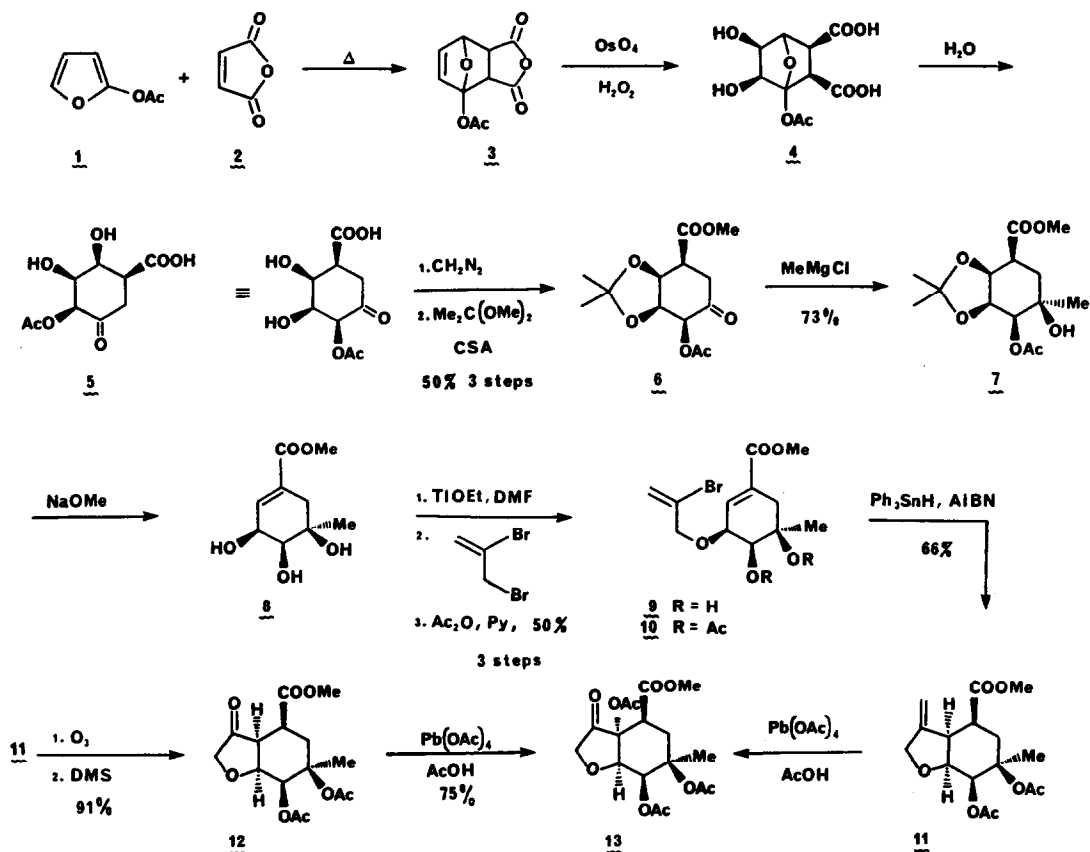


A Diels-Alder reaction between 2-acetoxifuran and maleic anhydride<sup>9</sup> led to the adduct 3, which was hydroxylated to the known diacid 4<sup>10</sup> in excellent overall yield. Upon stirring the diacid 4 in water, a remarkable transformation took place wherein the keto acid 5 was now formed<sup>11</sup>. Esterification gave the corresponding methyl ester, mp 154-155° (63%) which in turn was converted into the acetonide derivative 6, mp 149-150° (93%).<sup>12</sup> Treatment with methylmagnesium bromide in THF (5 min at 0°, total contact time) gave the adduct 7, mp 188-190° (72% based on recovered starting material).<sup>13</sup> Concomitant de-O-acetylation and  $\beta$ -elimination of the acetonide group upon treatment with sodium methoxide led to the triol 8, mp 200-203°, which was purified as its diacetate, mp 128-129.5° (50% overall from 7). At this juncture we explored a number of alkylation procedures with 2,3-dibromo-1-propene<sup>14</sup> (NaH, KH, Ag<sub>2</sub>O, etc) which were problematic. Successful alkylation took place when freshly prepared thallium ethoxide<sup>15</sup> in DMF was used, to give the desired ether 9 as an oil in 69% yield (Scheme III).

The free-radical<sup>16</sup> mediated intramolecular Michael cyclization, was successfully accomplished using triphenyltin hydride and AIBN. The structure of the product 11, mp 159-161°, (66%) was fully substantiated by 400 MHz <sup>1</sup>H n.m.r. spectroscopy and by decoupling experiments. We next had to address the all-important oxidative insertion of hydroxyl or its equivalent at the ring juncture.<sup>17</sup> Thus, ozonolysis of 11 led to the corresponding ketone 12 which upon treatment with lead tetraacetate in glacial acetic acid<sup>18</sup> at 75°C gave the desired  $\alpha$ -acetoxy oxahydrindane derivative 13 in 75% yield.

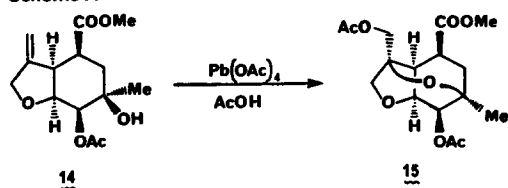
Surprisingly, treatment of 11 under the same conditions also led to the desired 13 in 30% yield, presumably via cleavage of a transient diol, followed by  $\alpha$ -acetoxylation of the resulting ketone. When the corresponding monoacetate 14 was subjected to the same reaction, the cyclic ether 15 was the major product. Inspection of molecular models reveal the proximity of the tertiary hydroxyl group to the activated exocyclic olefin in this cup-shape molecule (Scheme IV).

Scheme III

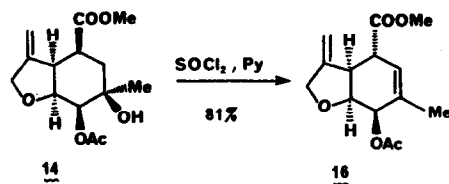


Unlike the published syntheses, the novel route described herein for the construction of functionalized oxahydrindanes leads to the 1-carbomethoxy derivatives which have their full complement of the required pattern of substitution in the intended subtarget. The tertiary hydroxyl group at C-4 (avermectin numbering) is a useful handle for the introduction of the endocyclic C-3/C-4 double bond at the opportune moment. We were successful in demonstrating the feasibility of such a regiospecific elimination by treatment of **11** with thionyl chloride in pyridine to give the oxahydrindene derivative **16** in 81% yield with concomitant epimerization at C-2 (Scheme V)<sup>19</sup>.

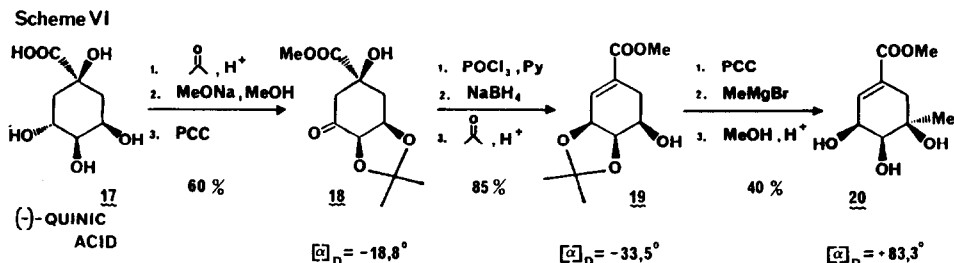
Scheme IV



Scheme V



Having worked out a viable synthetic route to racemic **13**, we next developed a synthesis of an enantiomerically pure advanced precursor from the readily available(-)-quinic acid **17** (Scheme VI). Thus, adapting the methodology already developed<sup>20</sup> to our needs, we were able to perform the transformations shown in Scheme VI, and to obtain the optically active triol **20**. The synthesis of the bicyclic ketone **13** in optically pure form should therefore be feasible according to the sequence developed in Scheme III.



**Acknowledgments.** We thank NSERC and FCAR for financial support and fellowships to Pierre Beaulieu and Daniel Dubé. We also thank Dr. Phan Viet Tan for <sup>1</sup>H n.m.r. measurements and decoupling experiments as well as Michael Evans for mass spectra.

#### References

- G. Albers-Schönberg, B.H. Arison, J.C. Chabala, A.W. Douglas, P. Eskala, M.H. Fisher, A. Lusi, M. Mrosik, J.L. Smith and R.L. Tolman, *J. Am. Chem. Soc.*, **103**, 4216 (1981).
- M. Mishima, M. Kurabayashi, C. Tamura, S. Sato, H. Kuwano and A. Saito, *Tetrahedron Lett.*, 711 (1975) and references cited therein.
- R. Baker, M.J. O'Mahony and C.J. Swain, *J.C.S. Chem. Comm.*, 1326 (1985); C. Yeats, D.A. Street, P. Kocienski and S.F. Campbell, *J.C.S. Chem. Comm.*, 1386, 1388 (1985); A.B. Smith, III, S.R. Schow, J.D. Bloom, A.S. Thompson and K.N. Winzenburg, *J. Am. Chem. Soc.*, **104**, 4015 (1982); D.R. Williams, B.A. Barner, K. Nishitani and J.G. Phillips, *J. Am. Chem. Soc.*, **104**, 4708 (1982).
- S. Hanessian, A. Ugolini, D. Dubé, P.J. Hodges and C. André, *J. Am. Chem. Soc.*, **108**, 2776 (1986).
- M.E. Jung and L.J. Street, *J. Am. Chem. Soc.*, **106**, 8327 (1984).
- M. Prashad and B. Fraser-Reid, *J. Org. Chem.*, **50**, 1565 (1985).
- A.P. Kozikowski and E. Maloney Huss, *Tetrahedron Lett.*, **26**, 5759 (1985).
- M.T. Crimmins and J.G. Lever, *Tetrahedron Lett.*, **27**, 291 (1986).
- N. Clauson-Kaas and N. Elming, *Acta Chem. Scand.*, **6**, 560 (1956);
- R. Daniels and J.L. Fischer, *J. Org. Chem.*, **28**, 320 (1963).
- G.E. McCasland, S. Furuta and L.J. Durham, *J. Org. Chem.*, **31**, 1516 (1966).
- All compounds were characterized by microanalytical data, mass spectrometry, <sup>1</sup>H n.m.r. (400 MHz) and other physical methods.
- Due to the propensity for elimination and other side-reactions, it was best to interrupt the reaction after a few minutes at 0°. Isolated yields 50-55%.
- R. Lespiau and M. Bourguel, *Org. Syn*, Coll. Vol. 1, 209 (1956); see also H. Nishiyama et al, *Tetrahedron Lett.*, **23**, 1267 (1982).
- H.O. Kalinowski, D. Seebach and G. Crass, *Angew. Chem. int. ed.*, **14**, 762 (1975).
- For some recent examples of radical-mediated carbocycle formation via vinyl halides, see N.N. Marinovic and H. Ramanathan, *Tetrahedron Lett.*, **24**, 1871 (1983); G. Stork and N.H. Baine, *J. Am. Chem. Soc.*, **104**, 2321 (1982); G. Stork and R. Mook, Jr., *J. Am. Chem. Soc.*, **105**, 3720 (1983); see also D.J. Hart, *Science*, **223**, 883 (1984) for a recent review of the general subject.
- Attempted allylic oxidation of **11** with selenium dioxide and t-butylhydroperoxide according to Sharpless (*J. Org. Chem.*, **47**, 2897 (1982)) led to the formation of the corresponding lactone.
- For examples in the steroid series, see T. Reichstein and C. Montigel, *Helv. J. Chim. Acta*, **22**, 1212 (1939); G.R. Chaudhry, T.G. Halsall and E.R.H. Jones, *J. Chem. Soc.*, 2725 (1961) and references cited therein.
- The identity of bicyclic structures was fully substantiated by detailed decoupling and N.O.E. studies at 400 Mz.
- G.A. Berchtold and D. Lesuisse, *J. Org. Chem.*, **50**, 888 (1985).

(Received in USA 27 June 1986)